



Department of Physical and Chemical Sciences

BACHELOR DEGREE IN CHEMICAL AND MATERIALS SCIENCES AND TECHNOLOGIES

**PROGRAMME OF “PHYSICS OF MATTER AND LABORATORY”**

**A.A. 2017-2018**

**Teacher:** Prof. Luca Lozzi; **ECTF:** 9

---

- **Elements of electromagnetism**
  - o Electric and Magnetic fields
  - o Maxwell equations
  - o Electromagnetic waves
  - o Interference and Diffraction
- **Special Theory of Relativity**
  - o The Galilean transformation
  - o The Michelson-Morley experiment
  - o Einstein's postulates
  - o The Lorentz transformation
  - o Relativistic dynamics
- **The crisis in Classical Physics: Particle-like properties of electromagnetic waves**
  - o Black-body radiation, stationary waves, Rayleigh-Jeans law.
  - o Quantisation of Energy and Planck theory
  - o The photoelectric effect
  - o The Compton effect
- **The structure of the atom**
  - o The Thomson's model
  - o The Rutherford's model
- **The electronic structure of the atoms**
  - o The Bohr's model, quantisation of the states, correspondence principle
  - o The Franck-Hertz experiment
- **Wave-like properties of particles**
  - o The de Broglie wavelength
  - o The wave packets
  - o Wave superposition and beats
  - o Bohr quantisation rules and de Broglie theory
  - o The Heisenberg principle of uncertainty

- Wave functions and probability amplitude
- Dual nature of particles and waves
- **The Schrodinger equation**
  - The Schrodinger equation and the time independent Schrodinger equation
  - Properties of the Schrodinger equation: eigenfunctions and eigenvalues
  - Operators in quantum mechanics
  - Examples of Schrodinger equations:
    - The free particle
    - Particle in 1-D box
    - Energy quantisation
    - 2-D and 3-D potential wells
    - The simple harmonic oscillator
    - The step potential
    - Barrier potential
    - The tunneling process
- **The hydrogen atom**
  - The Schrodinger equation for one-electron atoms
  - The Schrodinger equation in spherical coordinates
  - Solutions of the Schrodinger equation for one-electron atoms.
  - Radial and angular probability densities
  - The angular momentum operators
  - Magnetic dipole for the hydrogen atom
  - Magnetic dipole-magnetic field interaction, angular momentum precessing
  - The Stern-Gerlach experiment: the electron spin
  - Transition selection rules (outlines)
  - The Zeeman effect
  - The spin-orbit interaction (outlines)

**Laboratory sessions:**

- Determination of the prism refractory index using the Cauchy's law
- Determination of the Rydberg constant using a prism and a diffraction grating
- Determination of the electron e/m ratio
- The Stefan law on black body using an I-V circuit
- The photoelectric effect: determination of the Planck constant
- The Franck-Hertz experiment

Reading list:

1. K. Krane, Modern Physics, John Wiley & Sons (main text)
2. P.A. Tipler, Modern Physics, W.H. Freeman (same level of the main text)
3. F. Ciccacci, Fondamenti di Fisica Atomica e Quantistica, EdiSES (for some insights, only in Italian)
4. R. Eisberg, Fundamentals of Modern Physics, John Wiley & Sons (for some insights)
5. S. Gasiorowicz, Quantum Physics, John Wiley & Sons (only for students that want to study in depth quantum mechanics)
6. L. Colombo, Elementi di Struttura della Materia, Hoepli (only in Italian)

Assessment methods: written and oral examination